Tetrahedron: Asymmetry 20 (2009) 467-472

Contents lists available at ScienceDirect

Tetrahedron: Asymmetry

journal homepage: www.elsevier.com/locate/tetasy

Enzymatic resolution of 5-hydroxy- and 8-hydroxy-2-tetralols using immobilized lipases

Paolo Bonomi^a, Paola Cairoli^a, Daniela Ubiali^b, Carlo F. Morelli^a, Marco Filice^b, Ines Nieto^{b,c}, Massimo Pregnolato^b, Paolo Manitto^a, Marco Terreni^b, Giovanna Speranza^{a,*}

^a Dipartimento di Chimica Organica e Industriale, Università degli Studi di Milano and Italian Biocatalysis Center, via Venezian 21, 20133 Milano, Italy ^b Dipartimento di Chimica Farmaceutica, Pharmaceutical Biocatalysis Laboratories, Università degli Studi di Pavia and Italian Biocatalysis Center, via Taramelli 12, 27100 Pavia, Italy ^c Innovate Biotechnology srl and Italian Biocatalysis Center, Parco Scientifico Tecnologico, Strada Savonesa 9, 15050 Rivalta Scrivia (AL), Italy

ARTICLE INFO

Article history: Received 7 January 2009 Accepted 10 February 2009 Available online 13 March 2009

ABSTRACT

(R)-2-Tetralol (R)-**2a**, (R)-5-hydroxy-2-tetralol (R)-**2b** and (R)-8-hydroxy-2-tetralol (R)-**2c**, which are key intermediates in the synthesis of pharmacologically active 2-aminotetralins **3**, were prepared in moderate to very high enantiomeric excess (up to 99% ee) by enzymatic resolution of the corresponding racemic butyrates *rac*-**1a**, *rac*-**1b** and *rac*-**1c**, respectively, using lipases immobilized on octyl agarose. This methodology is an alternative to the microbial reduction of 2-tetralones.

© 2009 Elsevier Ltd. All rights reserved.

1. Introduction

2-Tetralols (1,2,3,4-tetrahydro-2-naphthols, **2**, Scheme 1) are useful intermediates in the synthesis of pharmacologically active 2-aminotetralins **3**.^{1,2} The 2-aminotetralin system represents the structural base of many dopaminergic, serotoninergic and adrenergic agents^{3,4} as well as compounds that interact with melatonin receptors.⁵ It has been found that the 2-aminotetralin activity is strongly dependent on both the substitution pattern of the aromatic ring and the configuration of the stereogenic centre at the 2-position.^{3,6,7} For example, 5-hydroxy-2-(*N*,*N*-di-*n*-propylamino)tetralin **3b** (5-OH–DPAT) (R₁, R₂ = *n*-Pr, Scheme 1) is a highly potent dopamine receptor agonist which has affinity for both the dopamine D₂ and the dopamine D₃ receptors,^{3,6,8} whereas the 8-hydroxy analogue **3c** (8-OH–DPAT) (R₁, R₂ = *n*-Pr) is a very

potent and selective serotonin 5-HT_{1A} receptor agonist, but it is devoid of dopaminergic activity.^{3,4,6,9} In the case of 5-OH–DPAT, the (*S*)-(–)-enantiomer is more active than the (*R*)-(+)-one.^{7,8} The opposite holds true for 8-OH–DPAT, with the (*R*)-(+)-enantiomer being the most active one.^{7,9} It should be noted that an affinity for the melatonin receptor requires a methoxy group at the 8-position of the 2-aminotetralin system and an amide instead of an amine moiety.⁵

2-Aminotetralin derivatives have found applications for the treatment of many central nervous system related disorders,^{5–7,10} and have been reported to possess a variety of actions including cardiovascular effects¹¹ and potent antifungal activities against pathogenic fungi.¹²

NR₁R₂

 $R_1, R_2 = H, alkyl$



2

lipase

OH

a X = H

b X = 5-OH **c** X = 8-OH





^{*} Corresponding author. Tel.: +39 02 5031 4097; fax: +39 02 5031 4072. *E-mail address:* giovanna.speranza@unimi.it (G. Speranza).

^{0957-4166/\$ -} see front matter @ 2009 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetasy.2009.02.011



Scheme 2. Reagents and conditions: (a) Pd/C (10%), dry *p*-cymene; (b) Na, EtOH reflux; then *p*-TsOH, EtOH-H₂O, 1:1; (c) NaBH₄, MeOH; (d) PhCH₂Br, K₂CO₃, CH₃CN; (e) butyryl chloride, TEA, DMAP, CH₂Cl₂; (f) H₂, Pd/C (10%), MeOH.

Despite the synthetic relevance of optically active 2-tetralols $\mathbf{2}$, their synthesis has been poorly investigated so far and mostly limited to the bioreduction of the corresponding 2-tetralones.^{13–21}

Recently, we found that the hydrolysis of racemic 2-tetralyl butyrate *rac*-**1a** catalyzed by *Rhizomucor miehei* lipase (*RML*) immobilized on octyl agarose affords (*R*)-alcohol (*R*)-**2a** with high enantiomeric excess as determined by MTPA ester method.²²

This prompted us to explore the possibility of extending such a methodology to prepare the synthetically useful 5-hydroxy- **2b** and 8-hydroxy-2-tetralols **2c** in enantiomerically pure form. Herein, we report that (R)-**2b** and (R)-**2c** can be obtained in moderate to very high enantiomeric excess (up to 99% ee) by lipase-catalyzed hydrolysis of the corresponding racemic butyrates *rac*-**1b** and *rac*-**1c**, respectively.

2. Results and discussion

2.1. Synthesis of substrates

5-Hydroxy-2-tetralyl butyrate *rac*-**1b** was prepared from 6-methoxy-1-tetralone **4b** as reported in Scheme 2, that is, by dehydrogenation over Pd/C in refluxing *p*-cymene, followed by Birch reduction of the intermediate 6-methoxy-1-naphthol **5b**.¹⁵ Further reduction of the resulting 5-hydroxy-2-tetralone **6b** with sodium borohydride gave racemic 5-hydroxy-2-tetralol *rac*-**2b** which was selectively protected at the phenolic hydroxyl group as a benzyl ether **7b** and then treated with butyryl chloride. Finally, the hydrogenolytic removal of the benzyl group from *rac*-**8b** afforded *rac*-**1b** in 80% yield from **6b**.

In the same way, 8-hydroxy-2-tetralyl butyrate *rac*-**1c** was synthesized starting from 7-methoxy-1-tetralone **4c** in 75% yield from **6c**. The ester *rac*-**1a** was prepared as previously reported.²²

2.2. Enzymatic resolution of 2-tetralols 2a-c

Six lipases from different biological sources (*Termomyces lanuginosus, Candida rugosa, Porcine pancreas, Pseudomonas fluorescens, Pseudomonas cepacia* and *R. miehei*), known for their hydrolytic activity, were immobilized on octyl agarose and tested for their enantioselectivity using 1-methylphenethyl butyrate *rac*-**9** as a 'model substrate'. The results of this preliminary screening are reported in Table $1.^{23}$ The enantioselectivity (*E*) was calculated according to Sih's method²⁴ from the reaction conversion and the enantiomeric excess of the product (ee_p). The extent of conversion was estimated by reversed phase HPLC, ee's by chiral HPLC, since the 1-phenyl-2-propanol **10** enantiomers could be easily separated on a OD-Chiralcel stationary phase. In all cases, the (*R*)-ester was preferentially hydrolyzed.

From these screening tests, lipases from *P. cepacia* (*PCL*), *P. fluorescens* (*PFL*) and *R. miehei* (*RML*) were selected and further evaluated as biocatalysts in the enantioselective hydrolysis of 2-tetralol butyryl esters *rac*-**1a**-**c**. Before submitting the racemic substrates to

Table 1

Screening of lipases immobilized on octyl agarose



Biological source of lipases	Immobilization yield ^b (%)	Activity ^c (U/g)	c ^d (%)	ee ^e _p (%)	E ^f
Termomyces lanuginosus	76	8	12	78 (R)	9
Candida rugosa	82	128	44	30 (R)	2
Porcine pancreas	84	38	50	2 (R)	1
Pseudomonas cepacia	98	40	46	99 (R)	>200
Pseudomonas fluorescens	96	47	50	98 (R)	>200
Rhizomucor miehei	40	917	49	99 (R)	>200

Hydrolysis of 1-methylphenethyl butyrate (*rac*-**9**) in MeCN–25 mM phosphate buffer, pH 7 (2:8) at 25 °C^a.

^a Substrate concentration, 2 mM.

^b Immobilization conditions: 42 mg of protein/g support, 25 °C.

^c The activity was measured at 25 °C towards a solution of ethyl butyrate (4 mL) in 100 mM phosphate buffer pH 7 (16 mL). The activity is expressed as U/g of support, U = μ mol of substrate hydrolyzed per minute.

 $d^{d}c$ (conversion), product % after 24 h.

^e Percentage of enantiomeric excess and absolute configuration of the alcohol **10** were determined by chiral HPLC, using (R)- and (S)-1-phenyl-2-propanol as reference samples.

^f E (enantioselectivity) = $\ln[1 - c(1 + e_p)]/\ln[(1 - c(1 - e_p)]; c, conversion.²⁴$

enzymatic resolution under the experimental conditions used for *rac*-**9**, we investigated the influence of the organic solvent on the catalytic properties of the immobilized lipases by incubating the enzyme preparations in the presence of different concentrations of acetonitrile in 25 mM phosphate buffer at pH 7 and 25 °C for 24 h. As it can be seen from Figure 1, increasing amount of acetonitrile results in a decrease of the lipase activities towards ethyl butyrate. In the case of *PFL* and *RML*, at 30% MeCN the enzymes lose their activities, whereas the immobilized *PCL* appears to be the most stable one considering that it retains a half of its maximum activity at 40% MeCN. On the basis of these data, the reaction conditions reported in Table 1 were chosen to test the enzymatic hydrolysis of *rac*-**1a**-**c**.



Figure 1. Residual hydrolytic activity towards ethyl butyrate of immobilized lipases in 25 mM phosphate buffer (pH 7) containing increasing amounts of MeCN after 24 h (see Section 4 for details).

Since neither 2-tetralols *rac*-**2a**-**c** nor their butyrates *rac*-**1a**-**c** were separated into enantiomers using the chiral HPLC column reported above for **10**, a number of chiral stationary phases were screened.²⁵ We succeeded in separating 2-tetralol *rac*-**2a** and 5-hy-droxy-2-tetralol *rac*-**2b** as well as its butyrate *rac*-**1b**, but not 8-hy-droxy-2-tetralol *rac*-**2c** on an S,S-WhelkO1 column. However, this column could be used to resolve 8-hydroxy-2-tetralyl butyrate *rac*-**1c** into enantiomers. This fact brought about the conversion of 8-hydroxy-2-tetralol, obtained by enzymatic hydrolysis, into its butyrate.

Thus, the absolute configuration of the alcohols obtained in the lipase-catalyzed hydrolysis of esters *rac*-**1a**-**c** was assigned by chiral HPLC comparison with authentic samples of (*S*)-2-tetralol (*S*)-**2a**, (*S*)-5-hydroxy-2-tetralol (*S*)-**2b** and (*S*)-8-hydroxy-2-tetralyl butyrate (*S*)-**1c**. Compounds (*S*)-**2a** and (*S*)-**2b** were obtained in 30% ee and 60% ee, respectively, by baker's yeast mediated reduction of the corresponding 2-tetralones.¹⁵ Analogous bioreduction of **6c** afforded (*S*)-**2c** (21% ee)¹⁵ which was converted into (*S*)-**1c** using the sequence of reactions outlined in Scheme 2.

The results of lipase-catalyzed hydrolysis of *rac*-**1a**-**c** (2 mM concentration) are summarized in Table 2.

Data of Table 2 indicate that the stereochemical course of resolutions performed by immobilized lipases is strongly dependent on both the biological source of the enzyme and the substitution pattern of the aromatic ring of the substrate. The most relevant result is the *PCL*-catalyzed hydrolysis of 8-hydroxy-2-tetralyl butyrate *rac*-**1c** to give the (*R*)-alcohol and the residual (*S*)-ester in almost enantiopure form (99% ee). Also the unsubstituted tetralol **2a** is resolved with good selectivity, particularly when *PFL* is used as the biocatalyst (92% ee). In contrast, a hydroxyl group at the 5-position seems to be scarcely tolerated and *rac*-**1b** was found to be a poor substrate. The best resolution (71% ee of the formed (*R*)-alcohol) was obtained using *PCL*. In all cases, the (*R*)-enantiomer is preferably hydrolyzed.

-			~
Ta	D	e	2

Enzymatic hydrolysis of esters 1a-c by lipases immobilized on octyl agarose^a

Substrate	Enzyme ^b	c ^c (%)	ee _p ^d (%)	E ^e
1a	PCL	46	75 (R)	13
	PFL	40	92 (R)	45
	RML	44	91 (R)	45
1b	PCL	46	71 (R)	11
	PFL	41	59 (R)	6
	RML	48	12 (R)	1
1c	PCL	50	99 (R)	>200
	PFL	50	87 (R)	41
	RML	50	52 (R)	5

 $^{\rm a}$ In MeCN–25 mM phosphate buffer pH 7 (2:8) at 25 °C, concentration of substrate 2 mM.

^b Pseudomonas cepacia lipase (PCL), Pseudomonas fluorescens lipase (PFL), Rhizomucor miehei lipase (RML); activity as in Table 1.

c (conversion), product % after 24 h.

^d Ee's and absolute configuration of (*R*)-**2a** and (*R*)-**2b** were determined by chiral HPLC of the reaction mixture on a S,S-WhelkO1 column; in the case of **2c**, its ee and absolute configuration were estimated via conversion into its ester (*R*)-**1c**.

^e *E* (enantioselectivity), see note of Table 1.

The enzymatic hydrolysis of 8-hydroxy-2-tetralyl butyrate *rac*-**1c** was scaled up to 20 mM using immobilized *PCL* as the biocatalyst in the presence of variable amounts of MeCN (Table 3). Under all the conditions employed, the alcohol (*R*)-**2c** and the remaining ester (*S*)-**1c** were obtained in almost enantiopure form (99% ee) and with excellent yield (45% and 43% isolated yield for (*R*)-**2c** and (*S*)-**1c**, respectively, at 20 mM concentration).

Table 3

Scale-up of PCL-catalyzed enzymatic hydrolysis of 1c in MeCN–25 mM phosphate buffer pH 7

Concentration of 1c (mM)	% MeCN	c ^a (%)	Time (h)	ee _p ^b (%)	ee _s d (%)	E ^e
2	20	50	6	99	99	>200
5	20	50	7	99	99	>200
5	30	50	8	99	99	>200
20	40	50	23	99	99	>200

^a % Conversion.

^b Enantiomeric excess of the (R)-**2c** was evaluated by chiral HPLC after conversion into the corresponding butyrate (R)-**1c** (an S,S-WhelkO1 column).

^d Enantiomeric excess of the remaining ester (S)-**1c** was evaluated by chiral HPLC.

^e *E* (enantioselectivity), see note of Table 1.

Also in the case of the 5-hydroxy isomer *rac*-**1b**, the enantioselectivity of the *PCL*-catalyzed hydrolysis was not affected by increasing amounts of acetonitrile. For example, at 20 mM concentration and in the presence of 40% MeCN, the isolated (*R*)-alcohol (*R*)-**2b** and (*S*)-ester (*S*)-**1b** were obtained with 71% ee and 65% ee, respectively (% conversion, 47; *E*, 11). A study of immobilization of *PCL* is under evaluation with the goal to highlight a possible influence of the matrix on the enantioselectivity. The results are forthcoming and will be disclosed in due course.

3. Conclusion

Optically active 2-tetralols (R)-**2b,c** that are useful intermediates in the synthesis of pharmacologically active 2-aminotetralins **3**, as well as their corresponding (S)-butyrates, have been prepared in high (or moderate) ee's by enantioselective hydrolysis of racemic esters *rac*-**1b,c** using lipases immobilized onto octyl agarose. Among the commercial lipases we tested, *PCL* appears to be the best biocatalyst for preparative purposes. It should be noted that the enzymatic resolution of *rac*-**2a–c** by properly selected lipases appears to be an alternative to microbial reductions of the corresponding 2-tetralones.^{13,15,19–21} In addition, the use of immobilized enzymes allows the easy recovery and the reuse of the biocatalyst with advantages in terms of product purification and process costs.

4. Experimental

4.1. General experimental

Analytical TLC was performed on silica gel F254 precoated aluminium sheets (0.2 mm layer, Merck, Darmstadt, Germany) using the following eluents: A. hexane/EtOAc (1:1): B. hexane/EtOAc (2:1): components were detected under an UV lamp and by spraving with a ceric sulfate/ammonium molybdate solution, followed by heating to ca. 150°. Silica gel 60, 40-63 µm (Merck) was used for flash chromatography. ¹H and ¹³C NMR spectra were recorded in CDCl₃ at 400.1 and 100.6, respectively, on a Bruker AVANCE 400 Spectrometer equipped with a XWIN-NMR software package (Bruker, Karlsruhe, Germany). Chemical shifts (δ) are given in ppm, and were referenced to the solvent signals ($\delta_{\rm H}$ 7.25, $\delta_{\rm C}$ 77.00 from TMS). EIMS spectra were run on a VG 7070 EQ mass spectrometer operating at 70 eV. A Merck-Hitachi L-7100 liquid chromatograph equipped with L-7300 column oven and a L-7400 UV detector was used for analytical HPLC. The OD-Chiralcel column was from Daicel® Chemical Industries Ltd; the S,S-WhelkO1 column was from Regis Technologies Inc. Optical rotations $(\alpha_{\rm D})$ were measured in MeOH at 25 °C on a Jasco P-1030 polarimeter.

All reagents were of commercial quality or purified prior to use by standard methods. Lipases from *T. lanuginosus* (TLL), *C. rugosa* (CRL), *P. pancreas* (*PPL*) were from Sigma–Aldrich (Milano, Italy). Lipases from *P. cepacia* (*PCL*) and *P. fluorescens* (*PFL*) were kindly donated by Amano Enzyme Inc. (Japan). *R. miehei* lipase (*RML*) (Chirazyme[®] L-9, sol.) was from Roche Diagnostics GmbH (Germany). Octyl agarose (Octyl Sepharose CL-4B) was from Sigma–Aldrich (Milano, Italy).

4.2. Synthesis of substrates

5-Hydroxy-2-tetralone **6b** and 8-hydroxy-2-tetralone **6c** were prepared from 6-methoxy- **4b** and 7-methoxy-1-tetralone **4c**, respectively, as described in Ref. 15. 1-Methylphenethyl butyrate **9** and 2-tetralyl butyrate **1a** were prepared as reported in Ref. 22.

4.2.1. Racemic 5-hydroxy-2-tetralol rac-2b

To a solution of 5-hydroxy-2-tetralone **6b** (1.35 g, 8.3 mmol) in anhydrous MeOH (100 mL) at 0 °C under N₂, NaBH₄ (600 mg, 15.9 mmol) was added portion wise. The reaction mixture was allowed to warm to rt, stirred for 3 h (TLC control) and then quenched by the addition of acetone (3 mL), water (100 mL) and a few drops of concd HCl. After removal of MeOH under reduced pressure, the remaining aqueous phase was extracted with EtOAc (3 × 100 ml). The organic extracts were dried (Na₂SO₄), evaporated under reduced pressure, and the residue was purified by flash chromatography (eluent A) to give racemic 5-hydroxy-2-tetralol *rac*-**2b** as a white solid (1.28 g, 7.8 mmol, 94% yield). TLC (eluent A), $R_f = 0.32$; ¹H NMR, ¹³C NMR and El MS as in Ref. 15.

4.2.2. Racemic 8-hydroxy-2-tetralol rac-2c

The title compound was prepared by NaBH₄ reduction of 8-hydroxy-2-tetralone **6c** in 91% yield as described above for 5-hydroxy-2-tetralol *rac*-**2b**. TLC (eluent A), $R_{\rm f}$ = 0.29; ¹H NMR, ¹³C NMR and El MS as in Ref. 15.

4.2.3. 5-Benzyloxy-2-tetralol rac-7b

To a stirred solution of 5-hydroxy-2-tetralol rac-2b (1.0 g, 6.1 mmol) in dry CH₃CN (100 mL), K₂CO₃ (840 mg, 6.1 mmol) and benzyl bromide (0.73 mL, 6.1 mmol) were added, and the reaction

mixture was heated at 70 °C for 24 h under an N₂ atmosphere. The solution was then cooled to rt and concentrated under reduced pressure. The resulting residue was triturated with Et₂O $(3 \times 100 \text{ mL})$, filtered and the filtrate was evaporated under reduced pressure. The residue was purified by flash chromatography (eluent B) to obtain pure 5-benzyloxy-2-tetralol rac-7b (1.43 g, 5.6 mmol, 92% yield). TLC (eluent A), $R_{\rm f} = 0.49$; ¹H NMR (CDCl₃) δ : 1.65 (s, 1H, OH), 1.78–1.88 (m, 1H, H-3_a), 2.09–2.12 (m, 1H, H-3_b), 2.74–2.83 (m, 2H, H-1_a, H-4_a), 3.00–3.13 (m, 2H, H-1_b, H-4_b), 4.14–4.20 (m, 1H, H-2), 5.09 (s, 2H, OCH₂), 6.75 (d, 1H, J = 7.6 Hz) and 6.77 (d, 1H, J = 8.0 Hz) (H-6 and H-8), 7.12 (app.t, 1H, J = 7.9 Hz, H-7), 7.33–7.47 (m, 5H, aromatic Hs); ¹³C NMR (CDCl₃) δ: 21.4 (CH₂), 31.1 (CH₂), 38.5 (CH₂), 67.1 (CH), 69.7 (CH₂), 108.6 (CH), 121.9 (CH), 125.0 (C), 126.4 (CH), 127.1 (CH), 127.8 (CH), 128.5 (CH), 135.8 (C), 137.4 (C), 156.3 (C); EI MS m/z (rel. int.) 254 (M⁺, 10), 236 (7), 162 (5), 144 (5), 91 (100).

4.2.4. 8-Benzyloxy-2-tetralol rac-7c

The title compound was prepared from 8-hydroxy-2-tetralol *rac*-**2c** in 88% yield as described for 5-benzyloxy-2-tetralol *rac*-**7b**. TLC (eluent A), $R_f = 0.55$; ¹H NMR (CDCl₃) δ : 1.61 (s, 1H, OH), 1.80–1.89 (m, 1H, H-3a), 2.02–2.09 (m, 1H, H–3_b), 2.67 (dd, 1H, J = 17.2, 7.8 Hz, H-1_a), 2.82–2.90 (m, 1H, H-4_a), 2.95–3.02 (m, 1H, H-4_b), 3.22 (dd, 1H, J = 17.2, 5.2 Hz, H-1_b), 4.16–4.22 (m, 1H, H-2), 5.09 (s, 2H, OCH₂), 6.76 (d, 1H, J = 8.1 Hz) and 6.77 (d, 1H, J = 7.9 Hz) (H-5 and H-7), 7.12 (app.t, 1H, J = 8.0 Hz, H-6), 7.33–7.48 (m, 5H, aromatic Hs); ¹³C NMR (CDCl₃) δ : 27.1 (CH₂), 31.0 (CH₂), 32.6 (CH₂), 67.3 (CH), 69.7 (CH₂), 108.4 (CH), 121.1 (CH), 123.6 (C), 126.3 (CH), 127.1 (CH), 127.8 (CH), 128.5 (CH), 137.2 (C), 137.4 (C), 156.7 (C); EI MS *m/z* (rel. int.) 254 (M⁺, 10), 162 (40), 91 (100).

4.2.5. 5-Benzyloxy-2-tetralyl butyrate rac-8b

A solution of 5-benzyloxy-2-tetralol rac-7b (510 mg, 2.0 mmol), triethylamine (0.30 mL, 2.1 mmol) and a catalytic amount of DMAP in dry CH₂Cl₂ (10 mL) was treated dropwise with butyryl chloride (0.3 mL 2.6 mmol). The reaction mixture was stirred at rt under an N_2 atmosphere until the complete disappearance of the starting material (TLC analysis, eluent B). Usual work-up afforded 5-benzyloxy-2-tetralyl butyrate rac-8b, pure by TLC, which was used for the next step without further purification (600 mg, 1.85 mmol, 93% yield). $R_{\rm f} = 0.76$ (eluent B); ¹H NMR (CDCl₃) δ : 0.97 (t, 3H, *I* = 7.3 Hz), 1.63–1.73 (m, 2H) and 2.31 (t, 2H, *I* = 7.5 Hz) (Hs of CH₃CH₂CH₂ moiety), 1.93–2.09 (m, 2H, H₂-3), 2.79–2.00 (m, 3H, H_2 -4 and H_{-1_a}), 3.14 (dd, 1H, J = 16.6, 4.80 Hz, H_{-1_b}), 5.10 (s, 2H, OCH₂), 5.21–5.26 (m, 1H, H-2), 6.74 (d, 1H, J = 8.2 Hz) and 6.77 (d, 1H, J = 7.7 Hz) (H-6 and H-8), 7.1 (app.t, 1H, J = 8.0 Hz, H-7), 7.35–7.48 (m, 5H, aromatic Hs); ¹³C NMR (CDCl₃) δ: 13.6 (CH₃), 18.5 (CH₂), 20.8 (CH₂), 27.4 (CH₂), 34.7 (CH₂), 36.5 (CH₂), 69.3 (CH), 69.8 (CH₂), 108.6 (CH), 121.7 (CH), 124.9 (C), 126.4 (CH), 127.1 (CH), 127.8 (CH), 128.5 (CH), 135.3 (C), 137.4 (C), 156.2 (C), 173.4 (C); EI MS m/z (rel. int.) 324 (M⁺, 1), 236 (20), 91 (100).

4.2.6. 8-Benzyloxy-2-tetralyl butyrate rac-8c

The title compound was prepared from 8-benzyloxy-2-tetralol *rac*-**7c** in 94% yield as described above for 5-benzyloxy-2-tetralyl butyrate *rac*-**8b**. R_f = 0.78 (eluent B); ¹H NMR (CDCl₃) δ : 0.94 (t, 3H, J = 7.1 Hz), 1.59–1.71 (m, 2H) and 2.25 (t, 2H, J = 7.3 Hz) (Hs of CH₃CH₂CH₂ moiety), 1.90–2.03 (m, 2H, H₂-3), 2.77 (dd, 1H, J = 17.5 and 6.6 Hz, H-1_a), 2.82–2.97 (m, 2H, H₂-4), 3.11 (dd, 1H, J = 17.5 and 5.4 Hz, H-1_b), 5.06 (s, 2H, OCH₂), 5.24–5.29 (m, 1H, H-2), 6.73 (d, 1H, J = 8.1 Hz) and 6.74 (d, 1H, J = 7.7 Hz) (H-5 and H-7), 7.12 (app.t, 1H, J = 7.9 Hz, H-6), 7.30–7.46 (m, 5H, aromatic Hs); ¹³C NMR (CDCl₃) δ : 13.4 (CH₃), 18.3 (CH₂), 26.3 (CH₂), 27.3 (CH₂), 28.8 (CH₂), 36.4 (CH₂), 69.1 (CH), 69.5 (CH₂), 108.2

(CH), 120.8 (CH), 122.9 (C), 126.1 (CH), 126.9 (CH), 127.6 (CH), 128.3 (CH), 136.9 (C), 137.1 (C), 156.4 (C), 173.1 (C); EI MS *m/z* (rel. int.) 236 (40), 145 (10), 91 (100).

4.2.7. 5-Hydroxy-2-tetralyl butyrate rac-1b

5-Benzyloxy-2-tetralyl butyrate rac-8b (500 mg, 1.5 mmol was dissolved in MeOH (40 mL) and hydrogenated in the presence of 10% Pd/C (320 mg) under 1 atm of H₂ at rt for 4 h (TLC control, eluent B). Filtration of the catalyst and removal of the solvent under reduced pressure furnished a residue, which was purified by passing through a short column of silica gel (eluent B) to give the desired product in quantitative yield (350 mg). TLC $R_{\rm f} = 0.59$ (eluent B); ¹H NMR (CDCl₃) δ : 0.98 (t, 3H, *I* = 7.4 Hz), 1.63–1.72 (m, 2H) and 2.31 (t, 2H, *I* = 7.3 Hz) (Hs of CH₃CH₂CH₂ moiety), 1.96-2.12 (m, 2H, H₂-3), 2.72-2.91 (m, 3H, H₂-4 and H-1_a), 3.10 (dd, 1H, J = 16.4, 4.8 Hz, H-1_b), 5.20-5.26 (m, 1H, H-2), 6.64 (d, 1H, J=8.0 Hz) and 6.70 (d, 1H, I = 8.0 Hz (H-6 and H-8), 7.03 (app.t, 1H, I = 8.0 Hz, H-7); ¹³C NMR (CDCl₃) *δ*: 14.3 (CH₃), 18.9 (CH₂), 20.7 (CH₂), 27.7 (CH₂), 35.0 (CH₂), 36.9 (CH₂), 69.6 (CH), 112.7 (CH), 122.0 (CH), 122.7 (C), 127.0 (CH), 135.9 (C), 153.6 (C), 173.9 (C); EI MS m/z (rel. int.) 234 (M⁺, 2), 163 (3), 147 (60), 146 (100).

4.2.8. 8-Hydroxy-2-tetralyl butyrate rac-1c

The title compound was prepared from 8-benzyloxy-2-tetralyl butyrate *rac*-**8c** in quantitative yield as described above for 8-hydroxy-2-tetralyl butyrate. TLC $R_{\rm f}$ = 0.57 (eluent B); ¹H NMR (CDCl₃) δ : 0.97 (t, 3H, *J* = 7.4 Hz), 1.64–1.73 (m, 2H) and 2.31 (t, 2H, *J* = 7.5 Hz) (Hs of CH₃CH₂CH₂ moiety), 1.93–2.05 (m, 2H, H₂-3), 2.72 (dd, 1H, *J* = 16.9, 6.9 Hz, H-1_a), 2.82–2.96 (m, 2H, H₂-4), 3.10 (dd, 1H, *J* = 16.9, 5.0 Hz, H-1_b), 5.25–5.31 (m, 1H, H-2), 6.63 (d, 1H, *J* = 8.0 Hz, H-7), 6.72 (d,1H, *J* = 8.0 Hz, H-5), 7.1 (app t, 1H, *J* = 8.0 Hz, H-6); ¹³C NMR (CDCl₃) δ : 13.7 (CH₃), 18.5 (CH₂), 26.5 (CH₂), 27.5 (CH₂), 28.7 (CH₂), 36.6 (CH₂), 69.5 (CH), 112.1 (CH), 120.7 (C), 120.8 (CH), 126.5 (CH), 137.5 (C), 153.7 (C), 173.7 (C); El MS *m/z* (rel. int.) 234 (M⁺, 2), 163 (3), 147 (80), 146 (100).

4.2.9. (*S*)-2-Tetralol (*S*)-2a, (*S*)-5-hydroxy-2-tetralol (*S*)-2b, (*S*)-8-hydroxy-2-tetralol (*S*)-2c and (*S*)-8-hydroxy-2-tetralyl butyrate (*S*)-1c

(S)-2-Tetralol (S)-**2a** (30% ee, $[\alpha]_D^{25} = -21.6 (c \ 0.5, MeOH)), (S)-5-hydroxy-2-tetralol (S)-$ **2b** ${60% ee, <math>[\alpha]_D^{25} = -42.8 (c \ 0.5, MeOH)} and (S)-8-hydroxy-2-tetralol (S)-$ **2c** ${21% ee, <math>[\alpha]_D^{25} = -19.3 (c \ 0.5, MeOH)} were prepared by baker's yeast reduction of 2-tetralone, 5-hydroxy-2-tetralone$ **6b**and 8-hydroxy-2-tetralone**6c**, respectively, according to Ref. 15.

(*S*)-8-Hydroxy-2-tetralyl butyrate (*S*)-1c $\{[\alpha]_D^{25} = -8.4 \text{ (c } 0.5, \text{MeOH)}\}$ was prepared from (*S*)-8-hydroxy-2-tetralol (*S*)-2c as reported above for the racemic ester.

4.3. Immobilization of lipases on octyl agarose

Immobilization of lipases on octyl agarose was performed as previously reported.²² The amount of immobilized enzyme was determined by Bradford assay.²⁶ The enzymatic activity of each enzyme preparation was determined by ethyl butyrate hydrolysis as previously described.²²

4.4. Enzyme stability assays

PCL, *PFL* and *RML* on octyl agarose were incubated in 25 mM phosphate buffer (pH 7.0) containing different concentrations of MeCN at 25 °C for 24 h. The residual activities of the enzymes were determined by ethyl butyrate hydrolysis²² and were compared with those of the enzyme preparations prior to incubation.

4.5. Enzymatic hydrolysis

4.5.1. General procedure for lipase-mediated resolution of 1methylphenethyl butyrate *rac*-9 and esters *rac*-1a-c

The enzyme preparation was added to a solution of the ester rac-9 and rac-1a-c in MeCN and 25 mM phosphate buffer pH 7 (20:80 v/v) and the reaction mixture was mechanically stirred at 25 °C. During the reaction, the pH was maintained constant by automatic titration (Metrohm 718 STAT Tritino) and the hydrolytic activity was evaluated by measuring the NaOH (100 mM) consumption (μ mol NaOH/min \times mL of enzyme preparation). The progress of the reaction was monitored both by automatic titrator and reversed HPLC analysis of the organic phase obtained by extracting 500 uL aliquots of the reaction mixture with EtOAc (800 uL). Chromatographic conditions: column. LiChroCART 250-4 select-B (Merck, Darmstadt, Germany): eluent, MeCN/H₂O (8:2): flow rate. 0.8 mL/min for the hydrolysis of ester rac-9. and 1.0 mL/min for the hydrolysis of esters *rac*-1a-c; UV detector, λ 220 nm; temperature, 25 °C; retention times: 1-methylphenethyl butyrate 9 Rt 10.3 min; 1-phenyl-2-propanol 10, Rt 3.4 min; 2-tetralyl butyrate **1a** R_t 11.2 min; 2-tetralol **2a**, R_t 3.7 min; 5-hydroxy-2-tetralyl butyrate **1b**, *R*_t 8.6 min; 5-hydroxy-2-tetralol **2b**, *R*_t 3.0 min; 8-hydroxy-2-tetralyl butyrate 1c, Rt 9.1 min; 8-hydroxy-2-tetralol (2c), Rt 3.1 min.

4.5.2. Determination of enantiomeric excess and absolute configuration assignment of the alcohols 10, 2a and 2b

The enantiomeric excesses of the products arising from the enzymatic hydrolysis of 1-methylphenethyl butyrate (*rac*-**9**) and racemic esters **1a**, **b** were determined by chiral HPLC analysis of the organic phase obtained by extracting 500 μ L aliquots of the reaction mixture with EtOAc (800 μ L).

1-Phenyl-2-propanol **10** enantiomers were resolved on a OD-Chiralcel column. Chromatographic conditions: eluent, hexane/isopropanol (99:1); flow rate, 1 mL/min; *T*, 25 °C; detector, λ = 220 nm; (*S*)-**10**, *R*_t 16.3 min and (*R*)-**10**, *R*_t 19.0 min.

Enantioseparation of 2-tetralols **2a,b** and of 5-hydroxy-2-tetralyl butyrate **1b** was carried out using a S,S-WhelkO1 column at 25 °C and the detector set at λ = 280 nm.

In the case of **2a**, the eluent was *iso*-PrOH/*n*-hexane (2:98) at a flow rate of 1 mL/min; retention times (R_t): (R)-**2a** 11.6 min, (S)-**2a** 13.1 min.

In the case of **1b** and **2b**, a mixture of EtOH/*n*-hexane (2:98) was used as eluent at a flow rate of 2 mL/min; retention times (R_t): 5-hydroxy-2-tetralol, (R)-**2b** 26.6 min, (S)-**2b** 30.1 min; 5-hydroxy-2-tetralyl butyrate, (R)-**1b** 18.1 min, (S)-**1b** 16.8 min.

The absolute configurations of the alcohols **10**, **2a** and **2b** obtained by hydrolysis of *rac*-**9**, *rac*-**1a** and *rac*-**1b**, respectively, were assigned using (R)- and (S)-1-phenyl-2-propanol (**10**), (S)-2-tetralol ((S)-**2a**) and (S)-5-hydroxy-2-tetralol ((S)-**2b**) as reference compounds.

4.5.3. Isolation of (*R*)-8-hydroxy-2-tetralol (*R*)-2c and determination of its enantiomeric excess and absolute configuration

Hydrolysis of the 8-hydroxy-2-tetralyl butyrate rac-1c was stopped by filtering off the enzyme at about 50% conversion. MeCN was removed under vacuum, and the resulting aqueous solution was extracted with EtOAc. The organic phase was dried over Na₂SO₄, evaporated under reduced pressure and the residue was separated by flash chromatography using *n*-hexane/EtOAc (4:1) as eluent to give (*R*)-8-hydroxy-2-tetralol (*R*)-**2c** and (*S*)-8-hydroxy-2-tetralyl butyrate (*S*)-**1c**. (*R*)-8-Hydroxy-2-tetralol was converted into its butyrate as reported above for the racemic ester. Its ee and absolute configuration was determined by chiral HPLC comparison with an authentic sample of (*S*)-**1c** (*S*,S-WhelkO1 column, chromatographic conditions: eluent, *n*-hexane/EtOH (98:2); flow rate, 1 mL/min; *T*, 25 °C; detector, λ = 280 nm, (*R*)-**1c**, *R*t 13.7 min, (*S*)-**1c**, *R*t 17.6 min).

When enzymatic hydrolysis of *rac*-1**c** was carried out on 20 mM scale using *PCL* as a biocatalyst, at 50% conversion (*R*)-8-hydroxy-2-tetralol (*R*)-2**c** {99% ee, $[\alpha]_D^{25} = +92.5$ (*c* 0.5, MeOH)} and (*S*)-8-hydroxy-2-tetralyl butyrate (*S*)-1**c** {99% ee, $[\alpha]_D^{25} = -39.2$ (*c* 0.5, MeOH)} were isolated in 45% and 43% yield, respectively.

Acknowledgements

We thank T. Szczerba and J. Kocergin (Regis technologies Inc., USA) for their support in chiral HPLC screening. We are grateful to L. Ferralasco (Mitsubishi Italia S.p.A, Italy) and Amano Enzyme Inc. (Japan) for their generous gift of PFL and PCL. A special thank to N. Palumbo e L. Ciccarelli for their help in the early stages of the work.

References

- 1. Orsini, F.; Sello, G.; Travaini, E.; Di Gennaro, P. Tetrahedron: Asymmetry 2002, 13, 253–259.
- Honda, T.; Fujii, A.; Inoue, K.; Yasohara, Y.; Itagaki, Y.; Maehara, K.; Takeda, T.; Ueda, Y. PCT Int. Appl. WO 2003046197, 2003.
- Homan, E. J.; Copinga, S.; Elfstrom, L.; Veen, T.; Hallema, J.-P.; Mohell, N.; Unelius, L.; Johansson, R.; Wikstrom, H. V.; Grol, C. J. *Bioorg. Med. Chem.* 1998, 6, 2111–2126. and references cited therein.
- 4. Dutta, A. K.; Reith, M. E. A.; Fei, X.-S. Bioorg. Med. Chem. 2002, 619-622.
- Copinga, S.; Tepper, P. G.; Grol, C. J.; Horn, A. S.; Dubocovich, M. L. J. Med. Chem. 1993, 36, 2891–2898.
- Rodenhuis, N.; Timmerman, W.; Wikstrom, H. V.; Dijkstra, D. Eur. J. Pharmacol. 2000, 394, 255–263.
- Yu, H.; Liu, Y.; Malmberg, Å.; Mohell, N.; Hacksell, U.; Lewander, T. Eur. J. Pharmacol. 1996, 303, 151–162.
- van Vliet, L. A.; Tepper, P. G.; Dijkstra, D.; Damsma, G.; Wikström, H.; Pugsley, T. A.; Akunne, H. C.; Heffner, T. G.; Glase, S. A.; Wise, L. D. J. Med. Chem. 1996, 39, 4233–4237.

- Arvidsson, L.-E.; Hacksell, U.; Johansson, A. M.; Nilsson, J. L. G.; Lindberg, P.; Sanchez, D.; Wikström, H.; Svensson, K.; Hiorth, S.; Carlsson, A. J. Med. Chem. 1984, 27, 45–51.
- For example, for the preventive treatment of Parkinson's disease: Scheller, D.; Dressen, F. PCT Int. Appl. WO 2005063238 A1 20050714, 2005; for the treatment of depression: Scheller, D.; Breidenbach, A.; Selve, N. PCT Int. Appl. WO 2005009425 A1 20050203, 2005; for treatment of various types of pain: Beyreuther, B.; Scheller, D.; Freitag, J.; Bianchine, J. PCT Int. Appl. WO 2007147556 A1 20071227, 2007.
- 11. Montanari, S.; Cavalleri, P.; Santangelo, F.; Marchini, F. PCT Int. Appl. WO 9838155 A1 19980903, **1998**.
- Yao, B.; Ji, H.; Cao, Y.; Zhou, Y.; Zhu, J.; Lü, J.; Li, Y.; Chen, J.; Zheng, C.; Jiang, Y.; Liang, R.; Tang, H. J. Med. Chem. 2007, 50, 293–5300.
- Bégué, J.-P.; Cerceau, C.; Dogbeavou, A.; Mathé, L.; Sicsic, S. J. Chem. Soc., Perkin Trans. J 1992, 3141–3144.
- 14. Martinez, G. R. Tetrahedron: Asymmetry **1995**, 6, 1491–1494.
- 15. Manitto, P.; Speranza, G.; Monti, D.; Fontana, G.; Panosetti, E. *Tetrahedron* **1995**, *51*, 11531–11546.
- Abalain, C.; Buisson, D.; Azerad, R. Tetrahedron: Asymmetry 1996, 7, 2983– 2996.
- Aina, G.; Nasini, G.; Vajna de Pava, O. J. Mol. Catal. B: Enzym. 2001, 11, 367–371.
 Mogi, M.; Fuji, K.; Node, M. Tetrahedron: Asymmetry 2004, 15, 3715–3717. and references cited therein.
- Yadav, J. S.; Reddy, G. S. K. K.; Sabitha, G.; Krishna, A. D.; Prasad, A. R.; Rahaman, H.-U.; Rao, K. V.; Rao, A. B. *Tetrahedron: Asymmetry* **2007**, *18*, 717–723.
- Lavandera, I.; Kern, A.; Ferreira-Silva, B.; Glieder, A.; de Wildeman, S.; Kroutil, W. J. Org. Chem. 2008, 73, 6003–6005.
- Lavandera, I.; Höller, B.; Kern, A.; Ellmer, U.; Glieder, A.; de Wildeman, S.; Kroutil, W. Tetrahedron: Asymmetry 2008, 19, 1954–1958.
- Nieto, I.; Rocchietti, S.; Ubiali, D.; Speranza, G.; Morelli, C. F.; Fuentes, I. E.; Alcantara, A. R.; Terreni, M. *Enzyme Microb. Technol.* 2005, 37, 514–520.
- 23. Similar results were obtained when the substrate was 1-methylphenethyl acetate, as previously reported: Ubiali, D.; Filice, M.; Moro, P.; Speranza, G.; Cairoli, P.; Szczerba, T.; Terreni, M. In 17th International Symposium on Chiral Discrimination, Parma, Italy, September 11–14, **2005**.
- 24. Chen, C.-S.; Fujimoto, Y.; Girdaukas, G.; Sih, C. J. J. Am. Chem. Soc. 1982, 104, 7294–7299.
- 25. Chiral selector screening was performed by Regis Technologies Inc., Morton Grove, IL 60053, USA (http://www.registech.com/). The following columns were tested: Whelk-O 1, ULMO, DACH-DNB, Pirkle 1-J, Beta-Gem, Alpha-Burke 2, Leucine, Phenylglycine.
- 26. Bradford, M. M. Anal. Biochem. 1976, 72, 248-254.